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Synthesis antioxidants and antimicrobial evaluation of new arylazo-triazene hybrids via coupling of aryl diazonium salts with phenolic triazenes

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Abstract: A new series of phenolic triazenes and bis(triazenes), incorporating piperidine was synthesized. Thus, for, 4-(piperidin-1-yldiazenyl)-phenol (3) was synthesized by coupling *p*-hydroxyphenyldiazonium chloride (2) with piperidine. The synthesis of phenolic arylazotriazene and arylazo(bis)triazene hybrids incorporating piperidine moiety was described. The antioxidant and antimicrobial activities of the synthesized new products were evaluated.

Keywords: Triazenes, piperidine, arylazotriazenearylazo(bis)triazene, diazonium salts

1.Introduction

As a result of their extensive range of biological and pharmacological activities [1-6], chemistry of triazenes and related compounds has been the subject of an extensive and increasing interest. Several triazenes receiving attention owing to their promising antibacterial [7, 8], analgesic [9], antiviral [10], antimalarial [11], antitumor and cytotoxic [4, 12] activity. Furthermore, in metal triazenide complexes synthesis, triazenes most often used as ligands [13-16], and in spectrophotometric determination of transition metals [17-19] and mercury [20] as reagents. On the other hand, phenolic compounds are important pharmacophores in the medicinal and pharmaceutical fields. Phenolic compounds for example, coumarins, flavonoids, tannins and phenolic acids possess a potential antioxidant activity [21 - 25], and a large number of these compounds possess antitumor, antiatherosclerotic, antibacterial, antimutagenic, anticarcinogenic, antiinflammatory, antiviral activities [26, 27].

From molecular design point of view, the combination of biologically active two pharmacophores or molecules is a well-known process for the drug-like molecules synthesis, which allows the synthesis of more potent agents. Accordingly, the present study has

focused on the synthesis of some new phenolic triazenes and bis(triazenes), incorporating a bioactive moiety, such piperidine, morpholine and piperazine as a structural unit. In addition, the synthesis of some new arylazo phenolic-triazene hybrids and triazene -Mannich base hybrids, of pharmaceutical interest, has been investigated. The new compounds might possess considerable synthetic and pharmaceutical interest.

N-Coupling of the suitable diazonium salt with either primary or secondary amine is considered to be the conventional route to the synthesis of triazenes. This reaction proceeds by a two-stage mechanism with a diazonium cation as intermediate state and is subject to general base catalysis. Molecules of solvent or of amine may act as catalyst [28-30].

2. Results and Discussion:

In the present study, the synthesis of a new series of phenolic triazenes incorporating a heterocyclic moiety has been achieved by treating *p*-hydroxyphenyl diazonium chloride (2) with the appropriate heterocyclic *sec*-amine. Therefore, *N*-coupling of 2 with piperidine lead to the formation of 4-(piperidin-1-yldiazenyl)phenol (3) in the following equation.

HO
$$\stackrel{+}{\sim}$$
 NH₂ $\stackrel{+}{\sim}$ NH₂ $\stackrel{+}{\sim}$ N=N CI $\stackrel{+}{\sim}$ N=N CI $\stackrel{+}{\sim}$ CH₃COONa / EtOH $\stackrel{+}{\sim}$ N=N-N (0-5°)

The formation of the phenolic triazene 3 is of particular interest, because a variety of compounds having a piperidine as a structural unit have been synthesized, because of their potential application in medicinal chemistry. The analytical and spectral data of compound 3

are consistent with its structure. The mass spectra of compound 3 revealed molecular ion peak at m/z = 205. The fragmentation pattern of compound 3, is depicted in the following Scheme.

3:
$$m/z = 205 (18) [M]^+$$
 $m/z = 121 (26)$
 $m/z = 188 (14) [M-OH]^+$
 $m/z = 188 (14) [M-OH]^+$
 $m/z = 104 (26)$

Mass fragmentation pattern of 3

The synthesis of arylazo compounds and their mode of pharmaceutical activity have been reviewed [31-34]. A variety of compounds having the arylazo moiety as a structural unit have been synthesized and studied with interest centered on their application as dyes and pigments [35, 36]. The literature survey reveals that the arylazo compounds exhibits outstanding biological activities such as, antibacterial [37] antidiabetics [38], antitumor [39] and cytotoxic [40] activities.

In the present work, the possible synthesis of some new arylazo phenolic-triazene hybrids incorporating piperidine moiety as a structural unit has been investigated. Accordingly, coupling of the appropriate diazonium chloride with 4-(piperidin-1-yldiazenyl)-phenol (3), afforded a series of 2-(aryldiazenyl)-4-(piperidin-1-yldiazenyl)-phenols (4a,b)

(Scheme 1). In line with this, coupling of bis(diazotized) *p*-phenylenediamine with 4-(piperidin-1-yldiazenyl)-phenol (**3**) in a molar ratio (1:2), led to the formation of 2,2'-(1,4-phenylenebis(diazene-2,1-diyl))bis(4-(piperidin-1-yldiazenyl)phenol) (**5**). A similar reaction of bis(diazotized)benzidine with **3** gave. 2 2'-(biphenyl-4 4'-diylbis(diazene-2 1-

reaction of bis(diazotized)benzidine with 3 gave 2,2'-(biphenyl-4,4'-diylbis(diazene-2,1-diyl))bis(4-(piperidin-1-yldiazenyl)phenol) (6) (Scheme 1). Formulation of compounds 4a,b was based on analytical and spectral data. The IR spectrum of 4b as an example exhibited a strong band at 3425 cm⁻¹. Their mass spectra indicated molecular ion peaks at m/z = 323 and 325, respectively, and mass fragmentation patterns which supported their structures. The mass fragmentation pattern of 4a, as an example, is depicted (Scheme 1A)

Scheme 1

$$m/z = 105 (40)$$

H₃C

 $m/z = 218 (26)$
 $m/z = 323(33) [M]^+$
 $m/z = 310(100) [M+2-CH_3]^+$
 $m/z = 295 (58) [M-N_2]^+$
 $m/z = 189 (28)$
 $m/z = 105 (40)$

Mass fragmentation pattern of 4a

Scheme 1 - A

The synthesis of 2,6-bis(arylazo)phenolic triazenes of the type **7a**, **b** and **8** incorporating a piperidine moiety as a structural unit as shown in (Scheme 2), has been achieved by coupling of the appropriate diazonium chloride with 4-(piperidin-1-yldiazenyl)-phenol (**3**) in a molar ratio (2:1). The mass spectra of compounds **7a**,

b and **8** showed molecular ion peaks at m/z = 441, 473 and 445, respectively. In addition, the unambiguous synthesis of **7a** by coupling of diazotized *p*-toluidine with **4a**, supported its assigned structure. The fragmentation pattern of **7a**, as an example, is depicted in Scheme 2-A

2 Ar-
$$N=N$$
 2 HO $N=N-N$ 3 2 HO $N=N-N$ 3 2 HO $N=N-N$ 3 2 HO $N=N-N$ 3 Parameters of 7a Mass fragmentation pattern of 7a

Scheme 2 - A

The coupling reaction of 2-(aryldiazenyl)-4-(piperidin-1-yldiazenyl)-phenols of the type **4** with diazonium ions is of particular interest,

because it allows the synthesis of a wide range of mixed 2,6-*bis*(arylazo)phenolic triazenes. For instance, coupling of 4-(piperidin-1-

yldiazenyl)-2-(*p*-tolyldiazenyl)phenol (**4a**) with aryl diazonium ions derived from *p*-aminophenol and *p*-anisidine afforded 2-(4-hydroxyphenyl)-diazenyl)-4-(piperidin-1-yldiazenyl)-6-(*p*-tolyldiazenyl)phenol (**9**) and 2-(4-methoxyphenyl) diazenyl)-4-(piperidin-1-yldiazenyl)-6-(*p*-tolyl diazenyl)phenol (**10**), respectively. The advantage of this reaction is further illustrated by the synthesis of 6,6'-(1,4-phenylenebis(diazene-2,1-diyl))bis(4-(piperidin-1-yldiazenyl)-2-(*p*-tolyldiazenyl)-

phenol) (11) and 6,6'-(biphenyl-4,4'-diylbis(diazene-2,1-diyl))bis(4-(piperidin-1-yldiazenyl)-2-(p-tolyldiazenyl)phenol) (12), by coupling of (4a) with *bis*(diazotized) p-phenylenediamine and *bis*(diazotized)-benzidine, as depicted in Scheme 3.

The mass spectra of compounds 9-12 indicated molecular ion peaks at m/z = 443, 457, 776 and 853, respectively. The mass fragmentation pattern of 11, as an example, is depicted (Scheme 3A).

Scheme 3

$$m/z = 664 (25)$$

$$m/z = 545 (98)$$

$$m/z = 455 (60)$$

$$N = N$$

$$N$$

Scheme 3 - A

3. Antioxidant activity:

The antioxidant activities of the synthesiszed new products (1-12) shown in (Table 1) were evaluated using 2,2'-azinobis-(3-

ethylbenzthiazoline-6-sulphonic acid) (ABTS) assay that is one of the most common radical scavenging assays.

Table 1. Antioxidant activities of compounds (1-12) using ABTS method

Compounds	Absorbance of samples	% inhibition
Control of ABTS	0.510	0
Ascorbic-acid	0.060	88.2%
3	0.068	86.7%
4a	0.175	65.7%
4b	0.093	81.8%
5	0.308	39.6%
6	0.416	18.4%
7a	0.267	47.6%
7b	0.160	68.6%
8	0.089	82.5%
9	0.291	42.9%
10	0.323	36.7%
11	0.353	30.8%
12	0.382	25.1%

From the evaluated products, compound 3 and 8 exihibited the highest inhibition value which is 86.7% and 82.5 respectavelly related to the presence of hydroxyl group and pipridiene group in both structures supported the antioxidant activty. On the other hand, the other evaluated products showed low to moderate inhibition values.

4. Antimicrobial activities:

The antimicrobial activities of compounds under investigation were screened according to the illustrated procedure in literature against *Gram-positive* bacterium (*Staphylococcus aureus*), *Gram-negative* bacterium (*Escherichia coli*) and one fungus (*Candida albicans*) using Ampicillin and Colitrimazole as standard drugs. The procedure which was used in determination is illustrated in details in literature [41-42]. The % activity index for the synthesized new compounds was calculated by the formula:

% Activity Index = $\frac{\text{Zone of inhibition by test compound (diametre)}}{\text{Zone of inhibition by standard (diametre)}} x 100$

Table 2. Diameters of inhibition zones (mm), antibacterial and antifungal activities of the new investigated compounds 1-12

	E. coli		S. aureus		C. Albicans	
Cpd.No.	Diameter of inhibition zone(mm)	% Activity index	Diameter of inhibition zone (mm)	% Activity index	Diameter of inhibition zone (mm)	% Activity index
3	19	76	23	95.8	22	81.5
4a	8	32.0	11	45.8	5	18.5
4b	13	52.0	16	66.7	15	55.5
5	NA		3	12.5	NA	
6	NA		NA		NA	
7a	4	16.0	7	29.2	3	11.1
7b	9	36.0	13	54.2	8	29.6
8	15	60.0	17	70.8	13	48.1
9	6	24.0	5	20.8	3	11.1
10	NA		NA		NA	
11	NA		NA		NA	
12	NA		NA		NA	
Ampicillin	26	100	24	100	NA	
Colitrimazole	NA		NA		27	100

NA: No Activity

The investigations of antimicrobial screening data which are shown in (Table 2) indicated that, compound 3 exihibited the

highest inhibition value which related to the presence of hydroxyl group and pipridiene group in its structure. On the other hand, most of newly prepared compounds exhibited fair to good antimicrobial activities compared to that of the standard drugs.

5. EXPERIMENTAL

Melting points were uncorrected measured by Stuart SMP10 electric melting point apparatus. Infrared spectra were measured on a Mattson 5000 FTIR spectrometer. The ¹H NMR spectra were recorded on JEOL 500 MHz NMR Unit Mansoura University in (*DMSO-d*₆) solution, with TMS as internal reference. Chemical shifts (δ) are measured in ppm and relative to TMS. The mass spectra were determined on a GC-MS QP-1000 EX Shimadzu instrument. Elemental analyses were performed by Carlo Erba 1108 instrument. The purity of the new synthesized compounds was established by thin layer chromatography (TLC) on silica gel plates, 0.25 nm, 60 GF 254 (Merck) with visualization by UV lamp. Compounds 5, 6, 11 and 12 are insoluble in common ¹H NMR solvents.

Synthesis of the triazene 3:

In 100 ml conical flask, prepare a solution of Piperidine (0.5 ml, 5 mmol) dissolved in ethanol (15 mL) and (1.36 g) sodium acetate CH₃COONa.3H₂O, cool the solution to 5° C by immersing in an ice bath. Stir the solution vigorously and add very slowly the cold diazonium salt solution prepared by dissolving 4-aminophenol (1) (0.55 g, 5 mmol) in Conc. HCl (1.5 mL, and 5 mL cold H₂O) and diazotized by addition of NaNO2 solution (0.36 g, 5.2 mmol; in 2 mL cold H₂O). When all diazonium salt solution has been added allow the reaction mixture to stand in an ice bath with occasional stirring for about 0.5 hour. The product was filtered, washed with cold water, dried, and crystallized from ethyl alcohol to give compound 3

4-(Piperidin-1-yldiazenyl)phenol (3).

Yield 50%, (brown powder), mp 151-152° C; IR (KBr): v = 3260 (OH), 2928 (-CH₂-aliphatic), 1598, 1511, 1400, 1332, 1102, 766 cm⁻¹; ¹H-NMR (DMSO-d6) $\delta = 1.15$ -1.22 (m, 6H, 3-H₂, 4-H₂, 5-H₂ of piperidine), 2.35-2.73 (m, 4H, 2-H₂, 6-H₂ of piperidine), 6.73-7.70 ppm (m, 4 H, Ar-H); 9.44 ppm (s, 1H, OH). MS (EI, 70 eV): m/z (%) = 205 (18) [M]⁺, 188 (14) [M-OH]⁺, 176 (23) [M-(N₂+1)]⁺, 160 (22) [M-(N₂+OH)]⁺, 104 (26) [M-(piperidinyl ion

+OH)]⁺, 121 (26) [M-piperidinyl ion]⁺, 93 (22) (HOC₆H₄-), 94 (73) [M+1-(piperidinyl ion+N₂)]⁺, 84 (22) [piperidinyl ion]⁺, 70 (100). *Anal.* Calcd for C₁₁H₁₅N₃O (205.26): Calcd. C, 64.37; H, 7.37; N, 20.47%. Found: C, 64.30; H, 7.30; N, 20.34%.

Synthesis of the arylazo-triazenes 4a, b. In 100 ml beaker, prepare a solution of triazene (3) (1.03 g, 5 mmol) in 30 ml 10 % sodium hydroxide solution, cool the solution to 5° C by immersing in an ice bath by direct addition of crushed ice. Stir the solution vigorously and add very slowly the cold diazonium salt solution prepared by dissolving 5 mmol of an appropriate aromatic amine (0.54 gm of ptoluidine for compound 4a and 0.55 gm paminophenol for compound 4b) in Conc. HCl (1.5 mL, and water 5 mL) and diazotized by addition of sodium nitrite solution (0.36 g, 5.2 mmol; in 2 mL cold H₂O). When all diazonium salt solution has been added allow the reaction mixture to stand in an ice bath with occasional stirring for about 1 hour. The product was filtered, washed with cold water, dried and crystallized from ethyl alcohol to give compounds 4a, and 4b.

4-(piperidin-1-yldiazenyl)-2-p-tolyldiazenyl)phenol (4a):

Yield 55%, (dark brown powder), mp 124- 126° C; IR (KBr): v = 3459 (OH), 2928 (-CH₂aliphatic), 1590, 1500, 1441, 1384, 1104, 821 cm⁻¹. 1 H-NMR (DMSO-*d6*) $\delta = 1.38-1.61$ (m, 6H, 3-H₂, 4-H₂, 5-H₂ of piperidine), 2.26-2.39 (m, 4H, 2-H₂, 6-H₂ of piperidine), 2.42 (s, 3H, CH₃), 6.89-7.76 (m, 7 H, Ar-H); 9.81 ppm (s, 1H, OH). MS (EI, 70 eV): m/z (%) = 323 (33) $[M]^+$, 310 (100) $[M+2-CH_3]^+$, 306 (20) $[M-CH_3]^+$ OH]⁺, 291 (32) [M-(CH₃+OH)]⁺, 218 (26) [M- $(CH_3C_6H_4N)]^+$, 208 (42)M+1- $(CH_3+OH+piperidinyl\ ion]^+$, 201 (28) [M- $(OH+CH_3C_6H_4N)]^+$ 189 (28) $[C_6H_4N=N$ piperidinyl+H]⁺, 112 (30) [N=N-piperidinyl]⁺, 105 (40) $[CH_3C_6H_4N]^+$, 104 (26) $[N=N-C_6H_4-$]⁺, 84 (37) [piperidinyl ion]⁺. Anal. Calcd for C₁₈H₂₁N₅O (323.40): Calcd. C, 66.85; H, 6.55; N, 21.66%. Found: C, 66.98; H, 6.50; N, 21.55%.

2-(4-hydroxyphenyl)diazenyl)-4-(piperidin-1-yldiazenyl)phenol (4b):

Yield 65%, (dark brown powder), mp 170-171°C; IR (KBr): v = 3452 (OH), 2930, 2856 (-

CH₂-aliphatic), 1590, 1500, 1499, 1446, 1383, 1269, 1129, 839 cm⁻¹. MS (EI, 70 eV): m/z (%) = 325 (31) [M]⁺, 297 (18) [M-N₂]⁺, 281 (22) [M+1-OH-N₂]⁺, 206 (16) [M+1-(HOC₆H₄N₂)]⁺, 196 (12) [M-(OH+N₂+piperidinyl)]⁺, 181 (12) [M+2-(2OH+N₂+piperidinyl ion]⁺, 124 (25) [[HOC₆H₄N₂+3H]⁺, 119 (15) [HOC₆H₃N=N-H]⁺, 107 (37) [HOC₆H₄-N]⁺, 104 (20) [N=N-C₆H₄-]⁺, 84 (10) [piperidinyl ion]⁺. *Anal.* Calcd for C₁₇H₁₉N₅O₂ (325.37): Calcd. C, 62.75; H, 5.89; N, 21.52%. Found: C, 62.61; H, 5.80; N, 21.41%.

2,2'-(1,4-phenylenebis(diazene-2,1-diyl)) bis(4-(piperidin-1-yldiazenyl)phenol) (5):

This compound was obtained following the procedure described above for compound 4 using (1.03 g, 5 mmol) of triazene (3) and pphenylenediamine (0.27 gm 2.5 mmol). Yield 50%, (deep brown powder), mp 126-127°C; IR (KBr): v = 3448 (OH), 2930, 2855 (-CH₂aliphatic), 1594, 1507, 1384, 1271, 1138, 840 cm⁻¹. MS (EI, 70 eV): m/z (%) = 542 (4) $[M+2]^+$, 541 (5) $[M+1]^+$, 540 (11) $[M]^+$, 458 (7) $[M+2-piperidinyl\ ion]^+$, 429 (21) $[M+1-N_2-piperidinyl\ ion]^+$ piperidinyl ion] $^+$, 335 (38) [M-(HOC₆H₄N=Npiperidinyl ion)]⁺, 316 (100) [M- (2 N=Npiperidinyl ion)] $^+$, 307 (47) [M-(HOC₆H₄N=N-(10)piperidinyl $[ion+N_2]^+$ 122 $[[HOC_6H_4N_2+H]^+, 119 (8) [HOC_6H_3N=N-H]^+,$ 112 (32) [2 N=N-piperidinyl ion)]⁺, 107 (37) $[HOC_6H_4-N]^+$, 105 (27) $[N=N-C_6H_5]^+$, 85 (8) [piperidinyl ion]⁺. Anal. Calcd for C₂₈H₃₂N₁₀O₂ (540.63): Calcd. C, 62.21; H, 5.97; N, 25.91%. Found: C, 62.04; H, 5.91; N, 25.80%.

2-(4'-(2-hydroxy-5-(piperidin-1-yldiazenyl)phenyl)diazenyl)-[1,1'-biphenyl]-4-yl)diazenyl)-4(piperidin1yldiazenyl)phenol (6)

This compound was prepared following the procedure described above for compound **4** using (1.03 g, 5 mmol) of triazene (**3**) and benzidine (0.46 gm 2.5 mmol). Yield 65%, (deep brown powder), mp 284-286° C; IR (KBr): v = 3451 (OH), 2934, 2854 (-CH₂-aliphatic), 1589, 1484, 1434, 1385, 1242, 826 cm⁻¹. ¹H-NMR (DMSO-d6) $\delta = 1.32$ -1.65 (m, 12H, 3-H₂, 4-H₂, 5-H₂ of piperidine), 2.22-2.37 (m, 8H, 2-H₂, 6-H₂ of piperidine), 6.85 (s, 2H Ar-H), 7.38-7.82 ppm (m, 14 H, 12 Ar-H + 2 OH); .MS (EI, 70 eV): m/z (%) = 618 (16) [M+2]⁺, 617 (24) [M+1]⁺, 616 (21) [M]⁺, 599

(27) [M-OH]⁺, 582 (18) [M-(2OH)]⁺, 536 (30) [M+4- piperidinyl ion]⁺, 514 (54) [M-1-OHpiperidinyl ion]⁺, 505 (22)[M-(N=Npiperidinyl ion)] $^{+}$, 498 (40) [M-(HOC₆H₄N=Npiperidinyl ion)] $^{+}$, 422 (18) [M+2-(N=Npiperidinyl ion)-piperidinyl ion]⁺, 182 (20) $[C_6H_5-C_6H_4-N=N]^+$, 152 (18) $[-C_6H_4-C_6H_4-]^+$, 120 (29) $[HOC_6H_3N=N-]^+$, 112 (12) [N=N-]piperidinyl ion)]⁺, 92 (30) [HOC₆H₃-]⁺, 84 (34) [piperidinyl ion]⁺. Anal. Calcd for C₃₄H₃₆N₁₀O₂ (616.73): Calcd. C, 66.22; H, 5.88; N, 22.71%. Found: C, 66.32; H, 5.95; N, 22.87%.

Synthesis of the bisarylazo-triazenes 7a, b and 8.

These compounds were prepared following the same procedure described above for compound 4 using triazene (3) (0.51 g, 2.5 mmol) and an appropriate aromatic amine 5 mmol (0.54 gm of *p*-toluidine for compound 7a, 0.62 gm of *p*-anisidine for compound 7b, and 0.55 gm *p*-aminophenol for compound 8).

4-(piperidin-1-yldiazenyl)-2,6-bis(-p-tolyldiazenyl)phenol 7a

Yield 60%, (red powder), mp 237-238 C; IR (KBr): v = 3448 (OH), 2921, 2860 (-CH₂aliphatic), 1606, 1510, 1449, 1382, 1171, 819 cm⁻¹. MS (EI, 70 eV): m/z (%) = 441 (56) [M]⁺, 426 (24) [M-CH₃]⁺, 411 (18) [M-(2CH₃)]⁺, 409 (32) $[M-(CH_3+OH)]^+,355$ (85)[M-2piperidinyl ion] $^{+}$, 352 (43) [M+2-(CH₃C₆H₄)] $^{+}$, 340 (42) [M-(piperidinyl ion +OH)]⁺, 335 (53) $[M-(CH_3C_6H_4+CH_3)]^+$ 333 (18)M- $(CH_3C_6H_4+OH)]^+$, 329 (65)[M-(piperidinylN=N)]⁺, 322 (65)M- $(CH_3C_6H_4N=N)]^+$ 313 M+1-(39) $(CH_3+piperidinyl N_2)]^+$, 314 (69)M-(piperidinylN=N+CH₃)]⁺, 231 (33)[M- $(CH_3C_6H_4N=N+CH_3C_6H_4-)$ ⁺,119 (57) $[CH_3C_6H_3N=N-]^+$, 112 (30) [N=N-piperidiny] $[ion]^+, 91 (28)$ $[CH_3C_6H_{4}-]^+$, 84 (30) [piperidinyl ion]⁺. Anal. Calcd for C₂₅H₂₇N₇O (441.54): Calcd. C, 68.01; H, 6.16; N, 22.21 Found: C, 68.10; H, 6.06; N, 22.30

4-(piperidin-1-yldiazenyl)-2,6-bis(4-methoxyphenyl)diazenyl)phenol (7b)

Yield 60%, (brown powder), mp 84-85° C; IR (KBr): v = 3455 (OH), 2930, 2875 (-CH₂-aliphatic), 1600, 1596, 1500, 1441, 1247, 1144, 839 cm⁻¹. ¹H-NMR (DMSO-d6) $\delta = 1.36-1.65$ (m, 6H, 3-H₂, 4-H₂, 5-H₂ of piperidine), 2.22-2.35 (m, 4H, 2-H₂, 6-H₂ of piperidine), 3.80 (s,

3H, OCH₃), 3.88 (s, 3H, OCH₃), 6.87 (s, 1 H, Ar-H); 6.89-6.91 (d, 2 H, Ar-H); 7.04 (s, 1 H, Ar-H);7.07-7.09 (d, 2 H, Ar-H); 7.72-7.74 (d, 2 H, Ar-H); 7.81-7.79 (d, 2 H, Ar-H); 9.94 ppm (s, 1H, OH). MS (EI, 70 eV): m/z (%) = 475 (4) $[M+2]^+$, 473 (28) $[M]^+$, 456 (22) $[M-2-(OH)]^+$, 442 (19) [M-(OCH₃)]⁺, 427 (25) [M+2- $(OCH_3+OH)]^+$ 394 (32)ſΜ- $(2OCH_3+OH)]^+,390$ (23)[M+1-(piperidinyl ion)] $^{+}$, 340 (16) [M+2-(CH₃OC₆H₄N=N)] $^{+}$, 335 (16) $[M-(CH_3OC_6H_4+CH_3O)]^+$,284 (38) [M+2-(CH₃OC₆H₄+piperidinyl ion)]⁺, 254 (27) [M- $(CH_3OC_6H_4N=N+piperidinyl\ ion)]^+\ 231\ (43)$ $[M-(CH_3OC_6H_4N=N+CH_3OC_6H_4-)]^+$, 136 (62) $[CH_3OC_6H_4N=N-]^+$, 112 (36) [N=N-piperidinyl ion)]⁺, 84 (40) [piperidinyl ion]⁺. Anal. Calcd for C₂₅H₂₇N₇O₃ (473.54) C, 63.41; H, 5.75; N, 20.71 Found: C, 63.50; H, 5.69; N, 20.65

4-(piperidin-1-yldiazenyl)-2,6-bis(4-hydroxyphenyl)diazenyl)phenol (8)

Yield 65%, (red powder), mp 108-109 C; IR (KBr): v = 3456 (OH), 2937 (-CH₂-aliphatic), 1610, 1590, 1499, 1447, 1383, 1245, 839 cm⁻¹. MS (EI, 70 eV): m/z (%) = 445 (22) [M]⁺, 417 $(25) [M-(N_2)]^+, 401 (27) [M+1-(N_2+OH)]^+, 363$ (54) [M+2-(piperidinyl ion)]⁺, 361 (15) [M-(piperidinyl $ion)]^+,333$ (45)[M-(N=Npiperidinyl ion)]⁺, 328 (48) [M+1-(piperidinyl ion+2OH)⁺, 324 (13)ſΜ-(19) $(HOC_6H_4N=N)]^+,315$ [M-1] $(OH+N=Npiperidinyl\ ion)$ ⁺, 306 (21) [M- $(HOC_6H_4N=N+OH)]^+$, 121 (35) $[HOC_6H_4N=N-$]⁺, 112 (18) [N=N-piperidinyl ion)]⁺, 84 (31) [piperidinyl ion]⁺. Anal. Calcd for C₂₃H₂₃N₇O₃ (445.48) C, 62.01; H, 5.20; N, 22.01 Found: C, 62.11; H, 5.10; N, 22.13

Synthesis of the bisarylazo-triazenes 7a, 9, 10, 11 and 12.

These compounds were prepared following the same procedure described above for compound 4 using 4-(piperidin-1-yldiazenyl)-2-p-tolyldiazenyl)phenol (4a) (0.81 g, 2.5 mmol) and an appropriate aromatic amine (0.27 gm (2.5 mmol)) of p-toluidine for compound 7a, (0.27 gm (2.5 mmol)) of p-aminophenol for compound 9, 0.31 gm (2.5 mmol)) of p-anisidine for compound 10, 0.14 gm (1.25 mmol)) of p-phenylenediamine for compound 11 and 0.23 gm (1.25 mmol) of benzidine for compound 12).

2-(4-hydroxyphenyl)diazenyl)-4-(piperidin-1-yldiazenyl)-6-(p-tolyldiazenyl)phenol (9)

Yield 75%, (red powder), mp 109-110°C; IR (KBr): v = 3449 (OH), 2929 (-CH₂-aliphatic), 1602, 1510, 1499, 1457, 1384, 1270, 835 cm⁻¹. ¹H-NMR (DMSO-*d6*) $\delta = 1.35$ -1.62 (m, 6H, 3-H₂, 4-H₂, 5-H₂ of piperidine), 2.19-2.37 (m, 4H, 2-H₂, 6-H₂ of piperidine), 2.45 (s, 3H, CH₃), 6.82-7.91 (m, 11H, 10 Ar-H + OH); 9.62 ppm (s, 1H, OH). MS (EI, 70 eV): m/z (%) = 443 (27) [M]⁺, 428 (19) [M-(CH₃)]⁺, 426 (17) [M-(OH)]⁺, 359 (17) [M-(piperidinyl ion)]⁺, 358 (22) [M-1-(piperidinyl ion)]⁺,352 (52) [M- $(CH_3C_6H_4)$, 331 (23) [M-(N=Npiperidinyl ion)] $^{+}$, 322 (33) [M-(HOC₆H₄N=N)] $^{+}$, 314 (36) $[M-1-(OH+N=Npiperidinyl ion)]^{+}309 (18) [M-1-(OH+N=Npiperidinyl ion)]^{+}309 (18)$ 1-(2OH+CH₃+piperidinyl ion)]⁺, 296 (100) [M-(2OH+N=Npiperidinyl ion)]⁺ 281 (21) [M-1- $(2OH+CH₃+piperidinylN=N ion)]^+$, 270 (35) [M+2-(CH₃C₆H₄+piperidinyl ion]⁺. Anal. Calcd for C₂₄H₂₅N₇O₂ (443.51) C, 65.00; H, 5.68; N, 22.11 Found: C, 65.11; H, 5.75; N, 22.05

2-(4-methoxyphenyl)diazenyl)-4-(piperidin-1-yldiazenyl)-6-(ptolyldiazenyl)phenol (10)

Yield 70%, (red powder), mp 89-90° C; IR (KBr): v = 3448 (OH), 2921 (-CH₂-aliphatic), 1605, 1510, 1449, 1382, 1270, 820 cm⁻¹. MS (EI, 70 eV): m/z (%) = 457 (12) [M]⁺, 427 (35) $[M+1-(OCH_3)]^+$, 413 (25)M+2- $(CH_3+OCH_3)^+$, 409 (12) $[M-(CH_3O+OH)]^+$, 359 (18) $[M+1-(CH_3+piperidinyl ion)]^+$, 356 $(17) [M-(OH+piperidinyl ion)]^+,350 (16) [M (CH_3OC_6H_4)$, 333 (11) $[M-(OH+CH_3OC_6H_4)]^+$, 324 (61) $[M+2-(CH_3O C_6H_4 N=N)]^+$,322 (23) $[M-(CH_3OC_6H_4N=N)]^+$ 321 (52)[M- $(OH+CH_3C_6H_4N=N)]^+,282$ (13)[M-1- $(CH_3C_6H_4+piperidinyl\ ion)]^+$, 279 (42) [M-1- $(HOC_6H_4+piperidinyl\ ion)]^+$, , 255 (26) [M- $(CH_3C_6H_4 + piperidinylN=N ion)]^+239 (22) [M-1]^+239 (22)$ (CH₃OC₆H₄ + piperidinylN=N ion)]⁺ 203 (100) $[M-1-(HOC_6H_2N=N-piperidinyl)]^+$, 270 (35) [M+2-(CH₃C₆H₄+piperidinyl ion]⁺. Anal. Calcd for C₂₅H₂₇N₇O₂ (457.54) C, 65.63; H, 5.95; N, 21.43 5.68; Found: C, 65.555; H, 5.94; N, 21.53

2-((4-(2-hydroxy-5-(piperidin-1-yldiazenyl)-3-(p-tolyldiazenyl)phenyl) diazenyl) phenyl) diazenyl)-4-piperidin-1-yldiazenyl)-6-p-tolyldiazenyl)phenol (11)

Yield 70%, (red powder), mp 260-261° C; IR (KBr): v = 3449 (OH), 2950 (-CH₂aliphatic), 1604, 1510, 1449, 1383, 1272, 820 cm⁻¹. MS (EI, 70 eV): m/z (%) = 776 (54) [M]⁺, $762 (50) [M+1-(CH_3)]^+, 760 (33) [M+1-(OH)]^+,$ 746 (42) $[M-2(CH_3)]^+$, 743 (79) $[M-1-(CH_3+$ (OH)⁺, 731 (53) $[M+2-(2CH_3+OH)]$ ⁺,692 (32) [M-(piperidinyl ion), 678 (33) (piperidinyl ion +CH₃)]⁺, 662 (25) [M-2-(piperidinyl ion N=N)]⁺,657 (46), $(CH_3C_6H_4N=N)]^+$, 608 (42) [M-2(piperidinyl ion)]⁺, 580 (33) [M-(piperidinyl+ piperidinyl N=N)⁺,545 (98) $[M-[M-(CH_3C_6H_4N=N+$ piperidinyl $N=N)]^{+}$ 538 (37) $2(CH_3C_6H_4N=N)]^+$ 455 (60)M+1- $(CH_3C_6H_4N=N)$ +HOC₆H₂N=Npiperidinyl)]⁺, $[M-(CH_3C_6H_4N=N$ 433 (46)piperidinylN=N)1⁺ 112 (24)[piperidinyl N=N)]⁺, 84 (28) [piperidinyl ion]⁺. Anal. Calcd for C₄₂H₄₄N₁₄O₂ (776.91) C, 64.93; H, 5.71; N, 25.24 Found: C, 64.85; H, 5.80; N, 25.33

2-(4'-((2-hydroxy-5-(piperidin-1-yldiazenyl)-3-(p-tolyldiazenyl)phenyl)diazenyl)-[1,1'biphenyl]-4-yl)diazenyl)-4-(piperidin-1yldiazenyl)-6-(p-tolyldiazenyl)phenol (12)Yield 72%, (red powder), mp 104-105° C; IR (KBr): v = 3450 (OH), 2934, 2853 (-CH₂aliphatic), 1599, 1484, 1434, 1355, 1286, 825 cm⁻¹. MS (EI, 70 eV): m/z (%) = 853 (59) [M]⁺, 769 (42) [M-(piperidinyl ion)]⁺, 745 (61) [M- $(CH_3C_6H_4+OH)]^+$ 728 (45)M- $(CH_3C_6H_4+2OH)]^+$, 710 (83) $[M-1-(2CH_3+$ piperidinyl ion)]⁺, 657 (58) [M-(piperidinyl + piperidinyl N=N)]⁺,649 (41) [M-1-(piperidinyl [M-2ion N=N+ $CH_3C_6H_4$), 627 (12)2(piperidinyl $N=N)]^{+}$ 612 (33)M-2-(piperidinyl N=N)+CH₃]⁺, N=N)]⁺. Anal. Calcd for C₄₈H₄₈N₁₄O₂ (853.01) C, 67.59; H, 5.67; N, 22.99 Found: C, 67.70; H, 5.75; N, 23.10

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